





Radiochemistry Webinars

Nuclear Fuel Cycle Series Nuclear Repository Science







In Cooperation with our University Partners





















Meet the Presenter...

Dr. Lindsay Shuller-Nickles

Dr. Lindsay Shuller-Nickles is an Assistant Professor in Environmental Engineering and Earth Science at Clemson University. She teaches undergraduate courses on subjects of mineralogy, petrology, the nuclear fuel cycle, and nuclear waste management and graduate courses on nuclear environmental engineering, technical nuclear forensics, and applications of quantum-mechanical modeling in environmental science. She received her Ph.D. in Materials Science and Engineering from the University of Michigan working with Rod Ewing and Udo Becker. Dr. Shuller-Nickles' research integrates computational and



experimental tools to gain a fundamental understanding of the behavior of radionuclide-containing materials in the environment. She currently supports three undergraduate students, four graduate students, and one post-doc working on two funded projects. The first, funded by the Department of Homeland Security, supports her research in nuclear forensics of the characterization of pre- and post-detonation solid materials. The second is an EPSCoR Implementation grant, which funds Dr. Shuller-Nickles' group as part of a much larger project (~\$5M for three years). Her work on the EPSCoR grant is focused on quantum-mechanical calculations to understand cation ordering, waste loading, and phase stability for advanced ceramic waste forms. The calculations are performed in collaboration with experimental efforts within the larger EPSCoR group.

Contact Information: lshulle@clemson.edu



OPERATED BY SAVANNAH RIVER NUCLEAR SOLUTIONS



Nuclear Repository Science

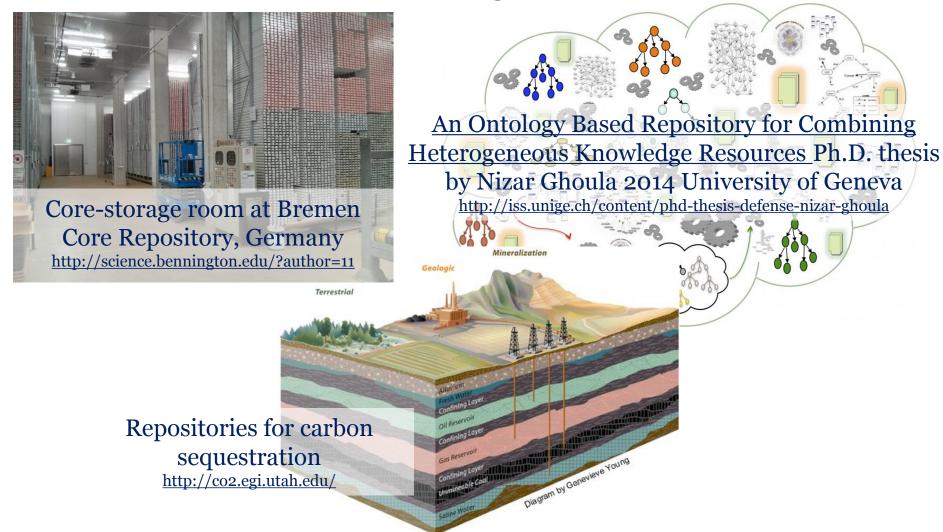
Dr. Lindsay Shuller-Nickles Clemson University



National Analytical Management Program (NAMP)

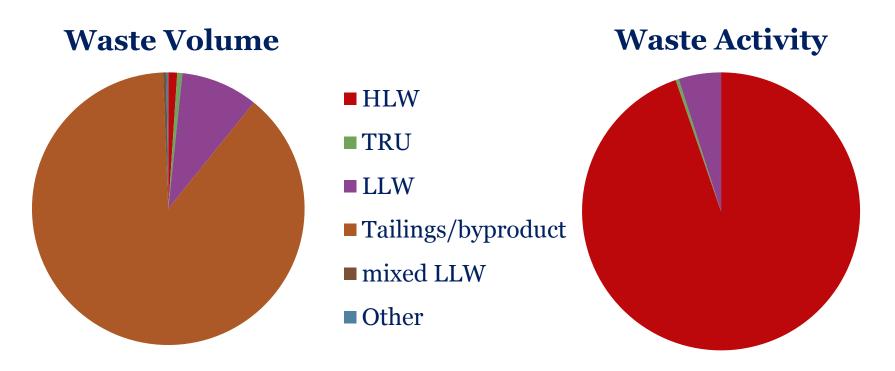
TRAINING AND EDUCATION SUBCOMMITTEE

Repository - a place where a large amount of something is stored



Why do we even need to understand nuclear repository science?

- Legacy waste (DOE managed)
 - 36 million m³ (1010 MCi)

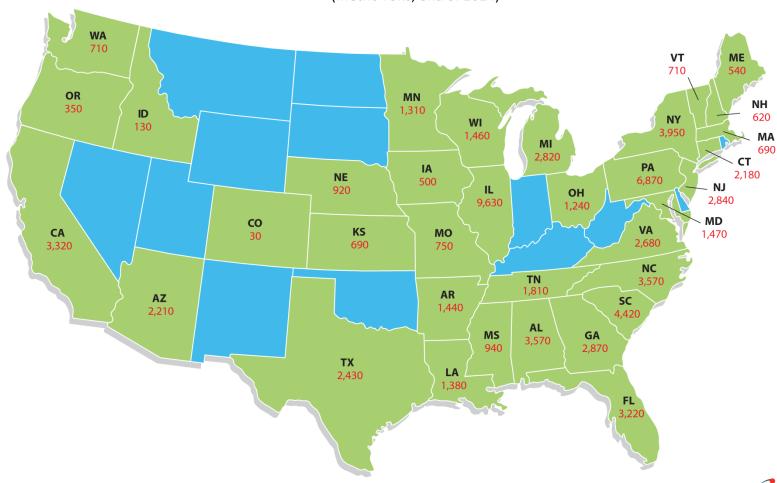


Why do we even need to understand nuclear repository science?

- Legacy waste (DOE managed)
 - 36 million m³ (1010 MCi)
- Global inventory¹
 - 300,000 MTHM total
 - 10,000 MTHM annual production
- US civilian inventory²
 - − ~72,000 MTHM total
 - − ~2,200 MTHM annual production

Used Nuclear Fuel in Storage

(Metric Tons, end of 2014)



Source: ACI Nuclear Energy Solutions



Long term disposal of radioactive waste

- 1957, Academy of Sciences report suggests
 - Underground storage as safest means for disposal.
 - Salt geology is best.
 - Scientific questions remain unanswered.

THE DISPOSAL OF RADIOACTIVE WASTE ON LAND

Report of the
Committee on Waste Disposal
of the
Division of Earth Sciences

Committee Members

Harry H. Hess, Chairman

John N. Adkins William B. Heroy
William E. Benson M. King Hubbert
John C. Frye Richard J. Russell
Charles V. Theis

Publication 519
Price \$1.00

National Academy of Sciences – National Research Council
Washington, D. C.
September 1957

- 2012, Blue Ribbon Commission's report suggests
 - Underground storage as safest means for disposal.
 - No specific site recommendations.
 - Scientific questions remain unanswered, but a sense of *urgency*

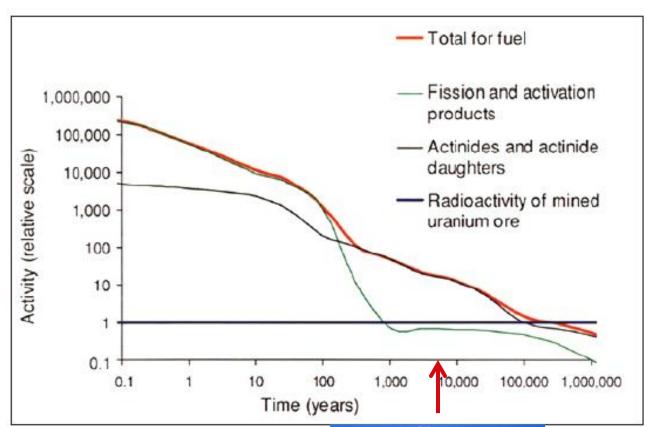


Report to the Secretary of Energy

JANUARY 2012 -



Why a geologic repository?



Relative radioactivity of SNF w/ a burn-up of 38 MWd/kg U. The activity is dominated by FP during the first 100 years, thereafter by actinides.

Pyramid of Djoser 27th century BC



Hedin *SKB Report* 1997; Bruno and Ewing *Elements* 2006

Health and safety requirements for disposal of HLW and SNF

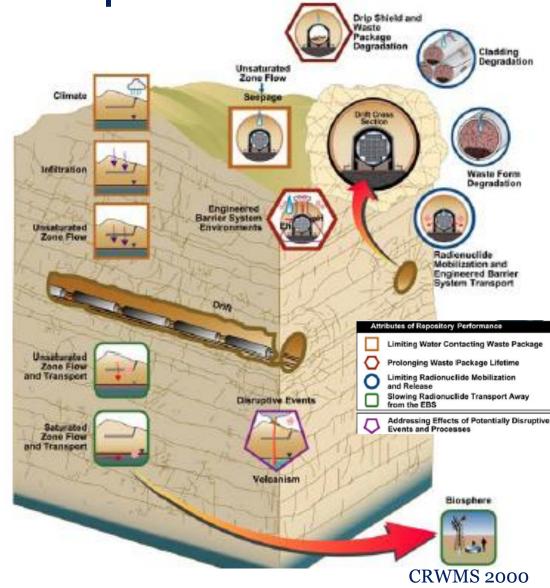
Country	Dose Constraint	Risk Limit	Compliance period
United States	0.15 mSv/yr 1.0 mSv/yr	Not specified Not specified	<10,000 yrs >10,000 yrs, but <1,000,000 yrs
Finland	Less than 0.1 mSv/yr. Release limits for various RNs established.	Not specified.	First several thousand yrs.
	Impacts should be comparable to those arising from natural radioactive materials but should remain insignificantly low.	Not specified.	Beyond first several thousand yrs.
France	0.25 mSv/yr for normal scenarios	Not specified	10,000 yrs
Sweden	Not specified	<10 ⁻⁵ /yr	100,000 yrs

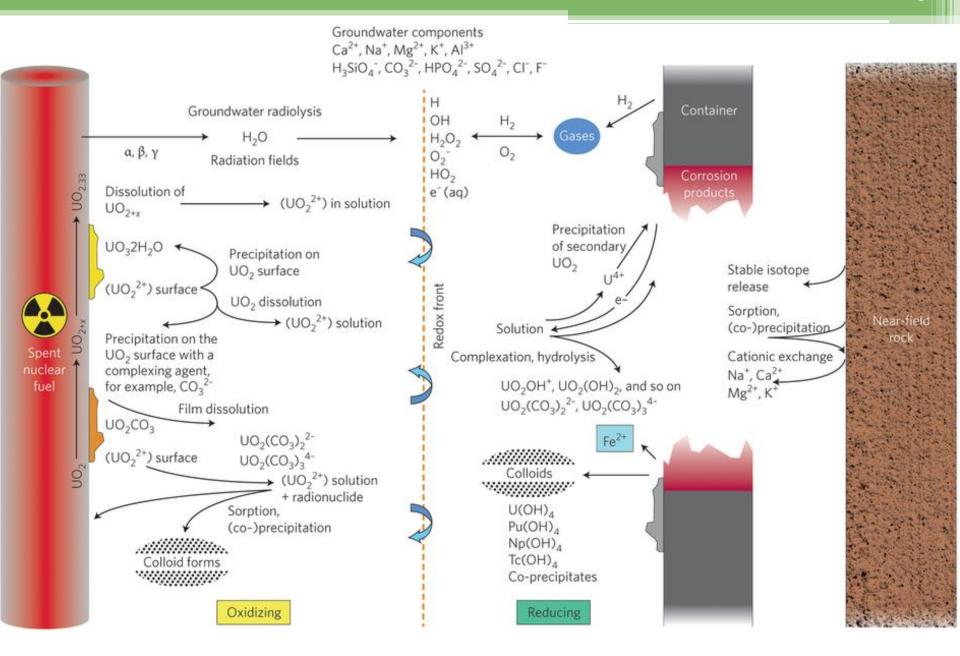
Demonstrating Compliance w/ Postclosure Standards

Country	Methodology
United States	Mean value of Monte Carlo realizations generated by a <u>probabilistic</u> Total System Performance Assessment
Finland	<u>Deterministic</u> , conservative safety case that addresses both the expected evolutions and unlikely disruptive events affecting long-term safety. The safety case consists of a numerical analysis based on experimental studies and will be complemented by <u>qualitative expert judgment</u> whenever quantitative analyses are not feasible or are too uncertain
France	Deterministic evaluation of several normal and altered scenarios. In addition, deterministic sensitivity calculations are used to evaluate the impact of uncertainty
Sweden	The regulations do not prescribe a specific methodology for demonstrating compliance. Both <u>deterministic and probabilistic</u> approaches can be used. Three types of scenarios are to be evaluated: (1) Main scenario – based on the probable evolution of the external conditions using realistic or pessimistic assumptions (2) Less probably scenarios – prepared for the evaluation of uncertainties. Include variations on the main scenario with alternative sequences of event. (3) Residual scenarios – include sequences of events and conditions that illustrate the significance of individual barriers and barrier functions.

YM TSPA Model Components

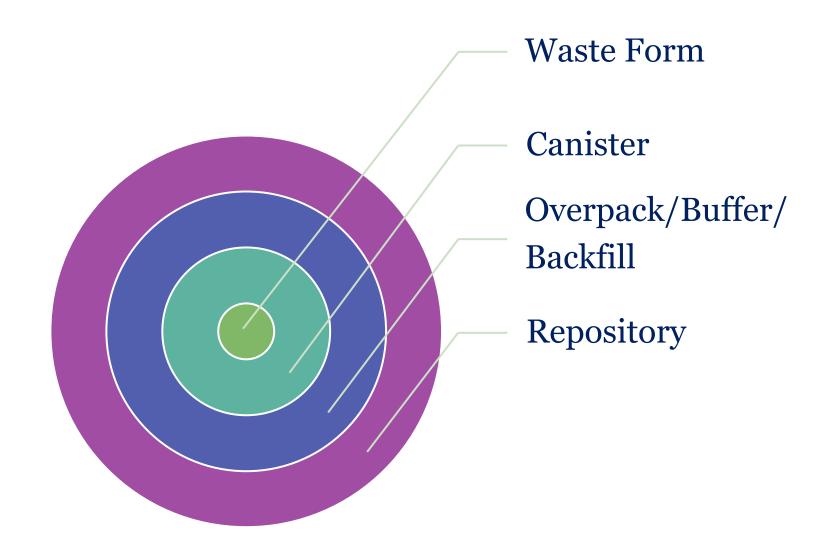
- Major model components are related to the attributes of the repository safety strategy
 - Seepage into Emplacement
 Drifts
 - Performance of Drip Shield
 - Performance of Waste Package Barriers
 - Solubility Limits of Dissolved
 Radionuclides
 - Retardation of Radionuclide Migration in UZ, SZ, and combinations thereof
- Natural and engineered barriers comprise the total system



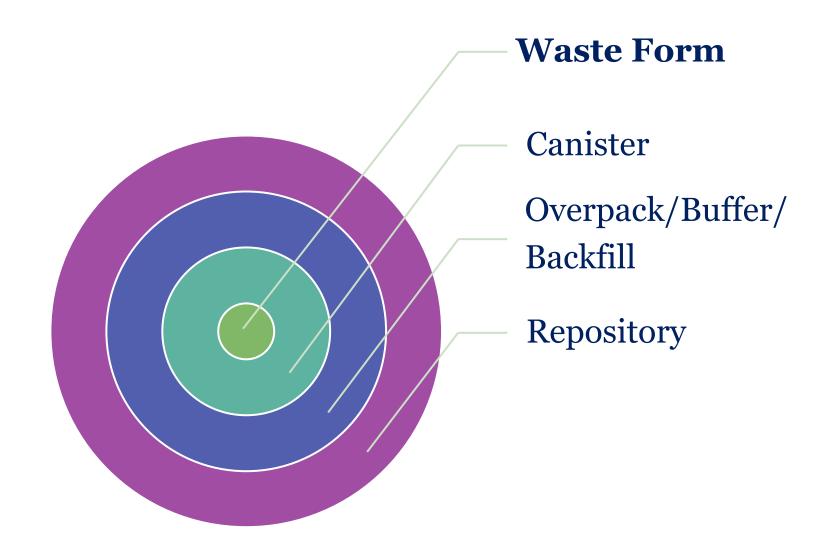


Ewing Nature Materials 2015

Multi-Barrier Approach



Multi-Barrier Approach



What is a Waste Form?

- Chemical form of the material holding the nuclear waste.
- Ideal waste form depends on the type of material being disposed.
 - -Specific activity, half-life, chemical speciation
- Long term disposal --> long lived isotopes
 - –If not reprocessing prior to disposal, some moderate half-life isotopes (*e.g.*, Cs-137, $t_{1/2}$ = 30.9 yr) remain in waste.

Examples of Waste Forms

- Spent nuclear fuel
- Borosilicate glass
- Ceramics (polycrystalline)
 - Synroc
 - Tailored ceramics
 - Glass-ceramics
 - Fuetap
- Ceramics (single phase)
 - titanate
 - zircon
 - pyrochlore

- "Novel" types
 - High silica porous glass matrix
 - Low-temperature hydroxylated ceramics
 - Clay & zeolite assemblages
- Multi-barrier
 - Coated particles (e.g., TRISO fuels)
 - Vitromet
 - Cermet

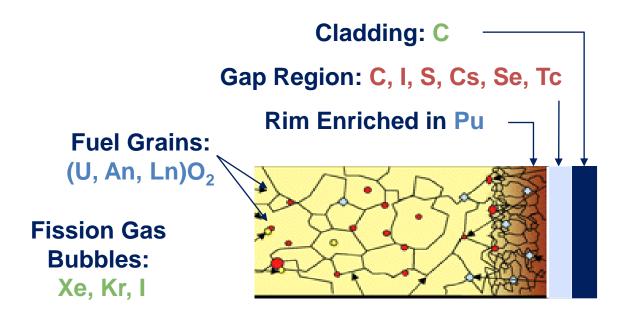
Principal Considerations for Waste Forms

- Chemical complexity of waste
- Large volumes of waste
- Ease of processing (and radiation safety)
- Durability (long-term) of waste form

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What is the chemical make-up of the waste?



Grain Boundaries:

Oxide Precipitates:

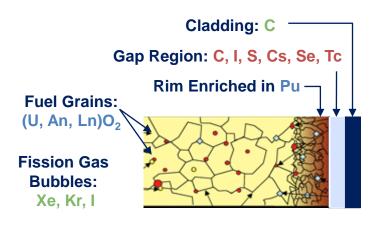
C, I, S, Cs, Se, Tc

Rb, Cs, Ba, Zr, Nb, Mo, Tc

ε-particles / Metallic Precipitates: Mo, Ru, Pd, Tc, Rh (Ag, Cd, In, Sn, Sb)

What is the chemical make-up of the waste?

Disposal Scheme	Direct Disposal	Reprocessing
Used fuel treatment	Cooling in wet and dry storage	Cooling followed by reprocessing
Waste form for disposal	Used nuclear fuel (oxide)	Engineered waste form (e.g., glass)

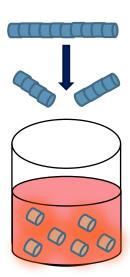


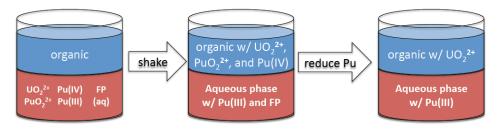
Grain Boundaries: Oxide Precipitates:

C, I, S, Cs, Se, Tc Rb, Cs, Ba, Zr, Nb, Mo, Tc

ε-particles / Metallic Precipitates:

Mo, Ru, Pd, Tc, Rh (Ag, Cd, In, Sn, Sb)

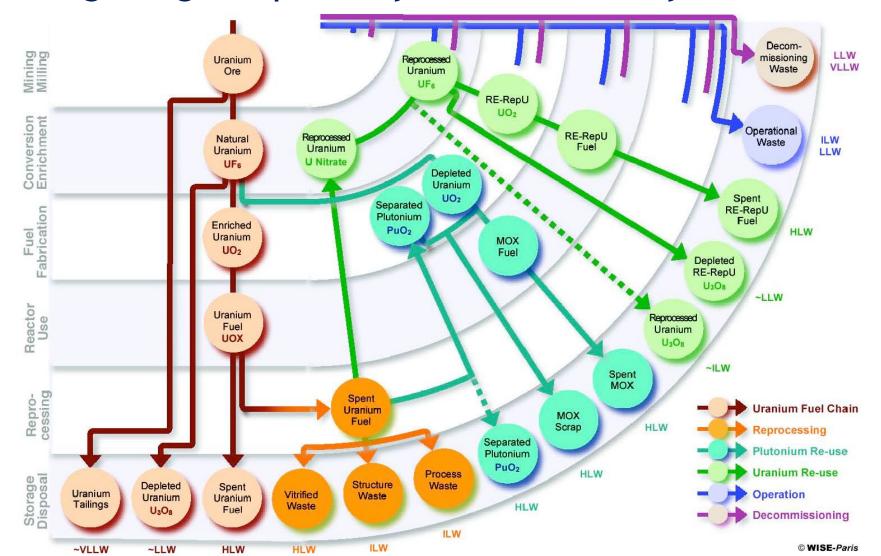




Aside: If we consider reprocessing, is this discussion of a geologic repository even necessary???

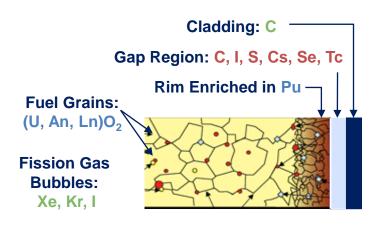
- Current Inventory: 72,000 MTHM
- US annual production: 2000 MTHM SNF
- La Hague annual capacity: 1700 tonnes per year¹
- More complex waste stream²

Aside: If we consider reprocessing, is this discussion of a geologic repository even necessary???



What is the chemical make-up of the waste?

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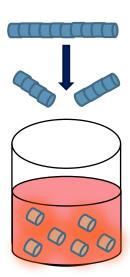
Grain Boundaries:

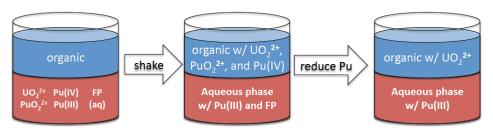
Oxide Precipitates:

C, I, S, Cs, Se, Tc Rb, Cs, Ba, Zr, Nb, Mo, Tc

ε-particles / Metallic Precipitates:

Mo, Ru, Pd, Tc, Rh (Ag, Cd, In, Sn, Sb)





Chemical Complexity - Fission Products

- Multi-phase waste forms
 - -Take advantage of complex waste stream
 - e.g., Where does the Mo go?
 - Glass soluble Cs-Mo-phase → release of Cs
 - Ceramic may incorporate into hollandite $(Ba_{x}Cs_{y})(A^{+3})_{2x+y}(Ti^{+4})_{8-2x-y}O_{16}$
 - Metallic alloys
 - Therefore...a multi-phase waste form (e.g., glass-ceramic) could help solve the Mo problem for glass waste forms.

Principal Considerations for Waste Forms

- Chemical complexity of waste
- Large volumes of waste
- Ease of processing (and radiation safety)
- Durability (long-term) of waste form

Waste loading

Waste form	Main phases	Application (waste loading)
Spent nuclear fuel	uraninite (~96%)	SNF from civilian NPP (100%)
Synroc-C	zirconolite, perovskite, hollandite, rutile	HLW from reprocessing (20%)
Synroc-D	zirconolite, perovskite, spinel, nepheline	US defense wastes (60-70%)
Pyrochlore	pyrochlore, zirconolite- 4M, brannerite, rutile	Separated actinides (35 wt%)
Monazite	monazite	Actinide-lanthinide wastes (25 wt%)
Borosilicate glass	glass, minor ceramics or soluble CsMo phase	Up to 20-30 wt%

Depends on chemical complexity, chemical compatibility, resistance to radiation damage

Principal Considerations for Waste Forms

- Chemical complexity of waste
- Large volumes of waste
- Ease of processing (and radiation safety)
- Durability (long-term) of waste form

TABLE 1 DATA ON NUCLEAR WASTE GLASS PRODUCTION. Until the year 2000, German waste was vitrified by AREVA NC (the French nuclear fuel cycle company, formerly COGEMA) and by British Nuclear Fuels, BNFL. Belgian, Japanese, and Swiss spent nuclear fuel waste is still transformed into glass at AREVA NC's plants. DWPF is the Defense Waste Processing Facility.

Vitrification plant	Location	Total waste glass produced (tonnes)	Number of canisters	TBq*
DWPF, Savannah River site	Aiken, South Carolina, USA	3600	2000	-
West Valley	New York, USA	_	300	-
BNFL	Sellafield, UK	900	2280	1 × 10 ⁷
AREVA NC	La Hague, France	4000	10,000	1.5 × 10 ⁸
Pamela	Mol, Belgium	500	2200	2.4 × 10 ⁵
Tokai Vitrification Facility	Japan	_	62	_

^{* 1} Tera-Becquerel (TBq) = 10¹² atoms decaying per second (or, for alpha and beta decay, transmutations/second)

Ease of Processing Demonstrated at the Defense Waste Processing Facility

- Projections (1981)
 - \$800 million
 - 1990 start-up
 - "...well established technology..."

- Actual
 - \$4 billion
 - 1996 start-up
 - "...first-of-a-kind technology that was well ahead of its time."

Wicks et al. 1993 Materials Research Society Bulletin

- "... the *excellent stability and technical performance* of waste glass forms and the ability of the glass... to retain radionuclides even when exposed to potential leachants within a repository environment."
- "These advantages fall into two general areas:
 - (a) ... good versatility, chemical durability, mechanical integrity, and radiation and thermal stability, and
 - (b) ease of fabrication ... well developed and demonstrated."

Principal Considerations for Waste Forms

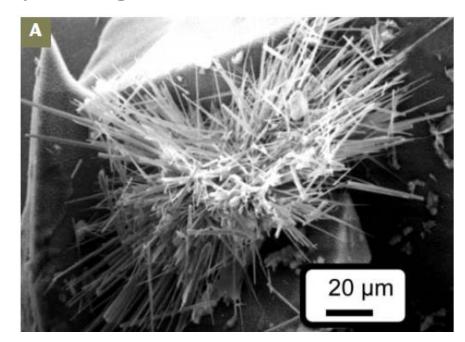
- Chemical complexity of waste
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Grambow (1994) Materials Research Society Bulletin

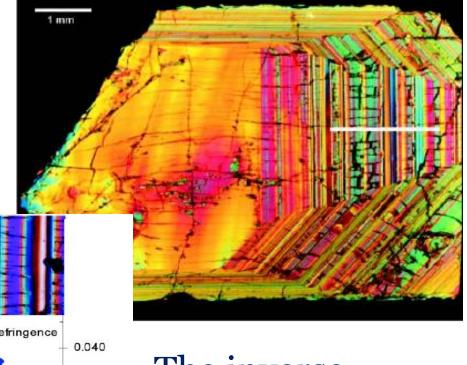
• "However, the glass corrodes slowly in water and humid air, and inevitably, certain quantities of radionuclides are mobilized. The glass is *not inherently corrosion-resistant*, but rather depends on the waste package and on surrounding geochemical and hydrological constraints."

Uranyl silicate mineral formed during leaching of AREVA NC-type glass in NaCl-rich brines

• Experiments performed at the Hahn Meitner Institute, Berlin at 110 °C, S/V = 2100 m⁻¹ for 831 days.



Zircon from Sri Lanka (560 Ma)

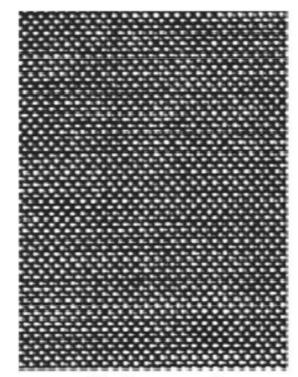


Birefringence 10 0.030 Jose (x 10 15 α/mg) 0.000 250 500 750 1000 1250 1500 1750 Position (µm)

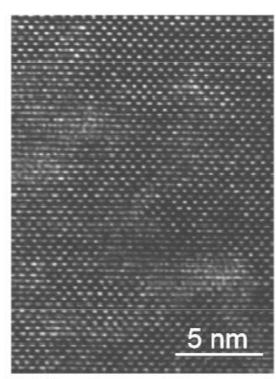
The inverse relationship between dose and birefringence as a function of position

Palenik et al. Am. Min. 2003

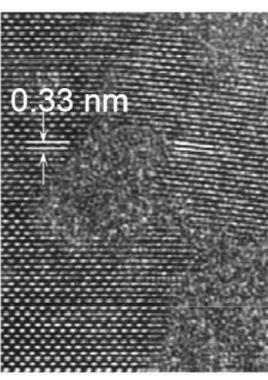
Radiation Damage in Natural Zircon from Alpha Decay and Ion Irradiation



Undamaged (no U or Th)

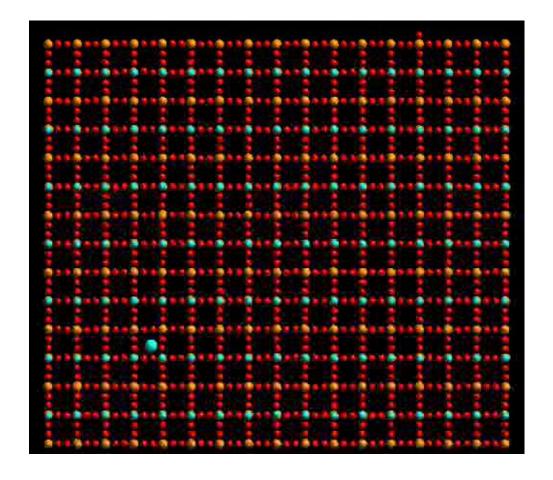


Alpha-Decay Damaged Over 550 Million Years From U & Th Decay

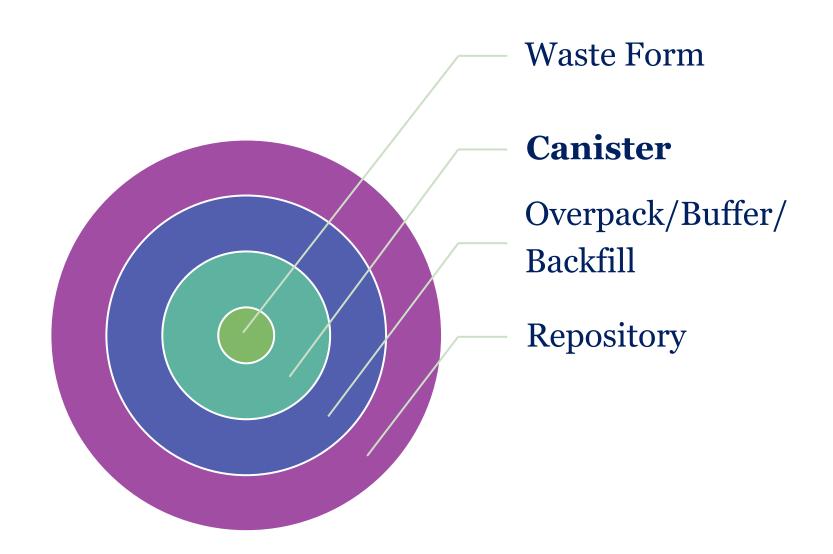


Ion Irradiation Damage 1.5 MeV Kr⁺ Ions

Molecular Dynamics Simulation of Atomic Collision Cascade: 30 keV U recoil in zircon



Multi-Barrier Approach



Canister functionality



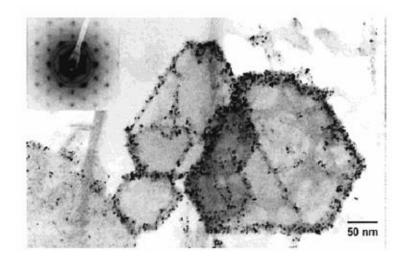
https://www.youtube.com/watch?v=1eJMY9MT4a8&feature=youtu.be&t=11

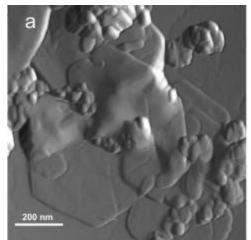
Candidate Canister Materials

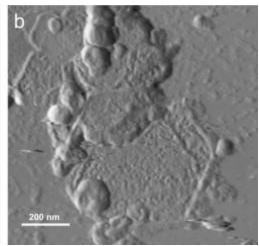
Country	Proposed Cani	ster material	Relevant alteration			
	outer canister	inner canister	phases			
United States	Alloy-22	Stainless steel	magnetite, maghemite,			
Belgium	Carbon steel *	Stainless steel	hematite, akaganeite,			
Japan	Carbon steel	Stainless steel	lepidocrocite, goethite,			
Switzerland	Cast iron	Stainless steel	siderite, green rust			
Finland	Copper	Cast-iron	reddish brown cuprous			
Sweden	Copper	Cast-iron	oxide, Cu ₂ O; green copper carbonates, sulfates, or oxychlorides; black copper sulfides			
* surrounded by thick concrete						

Fe-container as a Chemical Barrier: Formation of Green Rust [Fe(II)₄Fe(III)₂(OH)₁₂][CO₃/SO₄/OH·X H₂O]

 TEM image of U(VI) reaction with GR-OH • AFM showing Np reacted GR-Na,SO₄







O'Loughlin et al. ES&T 37 (2003) 721.

Christiansen et al. GCA 75 (2011) 1216.

Fe-container as a Chemical Barrier

Journal of Nuclear Materials 384 (2009) 130-139



Contents lists available at ScienceDirect

Journal of Nuclear Materials

journal homepage: www.elsevier.com/locate/jnucmat

UO₂ corrosion in an iron waste package

E.D.A. Ferriss a,*, K.B. Helean b, C.R. Bryan b, P.V. Brady b, R.C. Ewing a

^a Department of Geological Sciences, The University of Michigan, 2534 C.C. Little, 1100 N. University Avenue, Ann Arbor, MI 48109-1005, USA ^b Sandia National Laboratories, P.O. Box 5800, MS 0779, Albuquerque, NM 87185-0779, USA

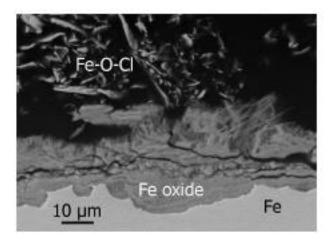


Figure 2.6. Back-scattered electron micrographs of Package D corroded steel of at 90 days. The polished cross-section shows oxidized areas along the steel surface and loosely consolidated fibers or plates of a Cl-rich phase.

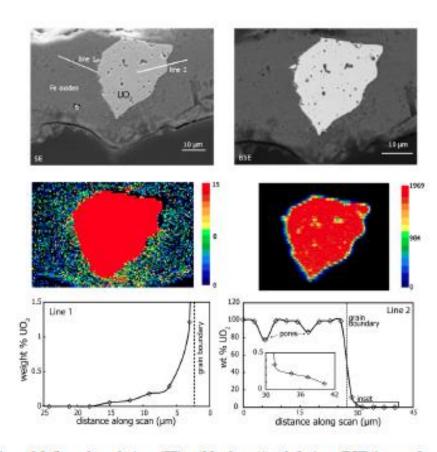
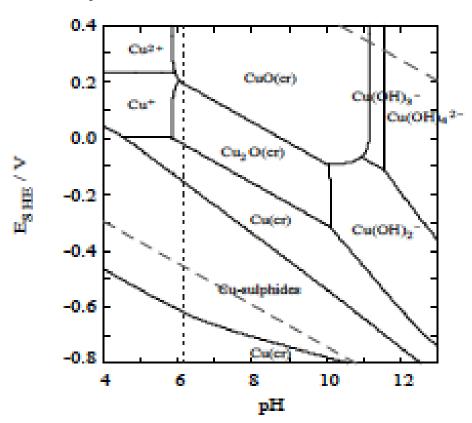


Figure 2.9. Secondary electron (SE) and back-scattered electron (BSE) image of UO₂ grain surrounded by steel corrosion products (most likely magnetite) in package E with associated EMPA/WDS line scans and elemental maps of U locused at different levels of total counts per pixel.

Cu-Container as a Chemical Barrier Estimated Lifetime > 100,000 years

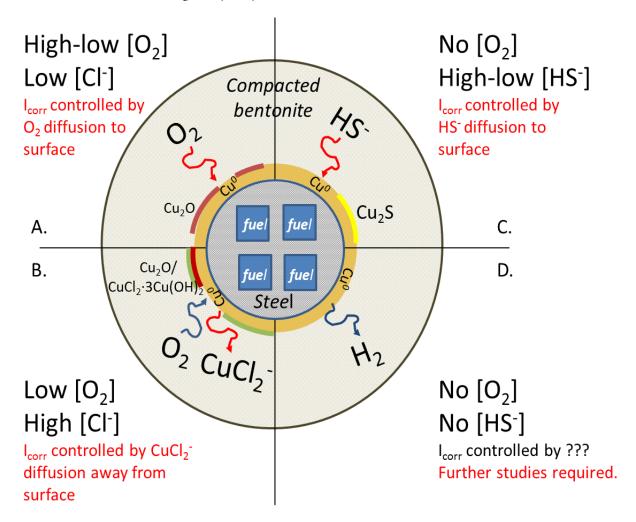




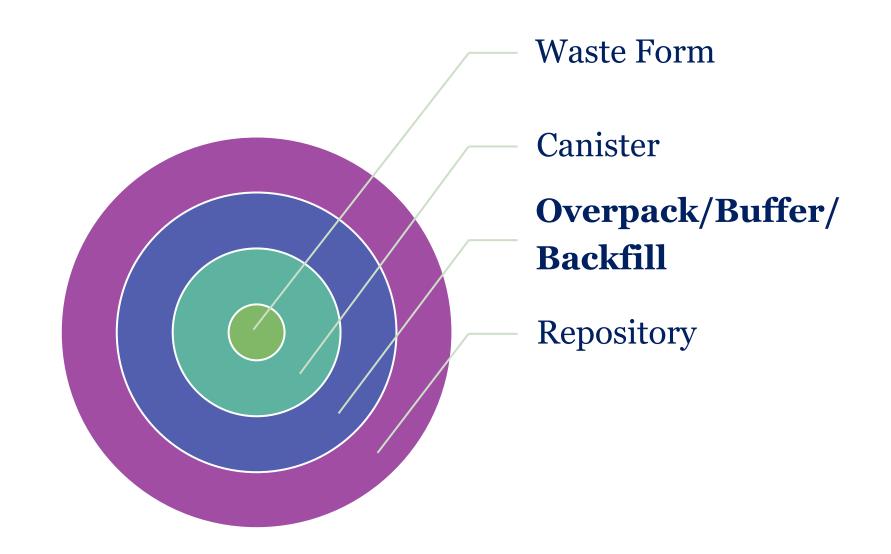
Pourbaix diagram for copper in solutions containing [HS⁻]_{TOT} = 0.2 mmol/kg & [Cu]_{TOT} = 10⁻⁶ mol/kg @ 100 °C

Puigdomenech et al. "Thermodynamic data for copper: Implications for the corrosion of copper under repository conditions" SKB 2000.

Cu case study(s)



Multi-Barrier Approach



Clay backfill

SOME ISSUES ON THE USE OF BACKFILL MATERIALS IN HIGH-LEVEL NUCLEAR

WASTE REPOSITORIES

Rex Couture

Argonne National Laboratory 9700 South Cass Avenue Argonne, Illinois 60439

July 1984

30 yearslater...there's still more to learn about backfill materials

Introduction

This manuscript describes critical issues suggested by the author regarding the use of clay backfill in repositories for high-level nuclear waste. The issues were identified from results in the NRC-sponsored research program, "Modification of Backfill Materials," Begg A2239, being conducted at Argonne National Laboratory.

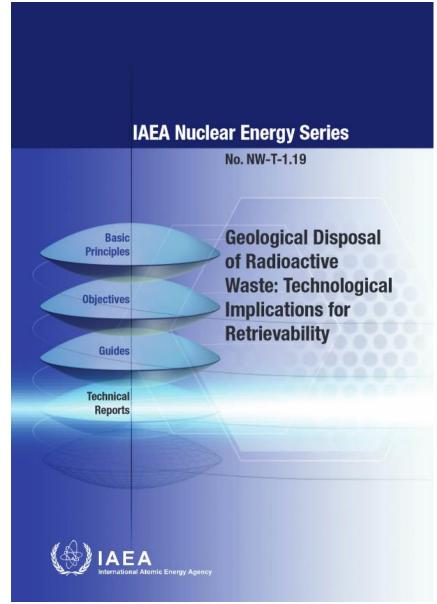
Our results from laboratory testing of proposed backfill sug that backfill is potentially very effective at limiting the flow groundwater past the waste package, and is therefore, very import essential to a basalt repository; it may also provide a necessary or safety for a tuff repository. I define backfill to mean a min barrier placed around the waste canisters. Begg J.D. et al. *J. of Env. Radioact.* 141 (2015) 106.

The aim of the current work is two-fold: to provide information on Pu sorption/desorption to/from industrial grade FEBEX bentonite, a potential repository backfill material, and to determine if the linearity observed for Pu(V) sorption to a pure Namontmorillonite (Begg et al., 2013) extends to Pu(IV) sorption to a multi-component clay rock material. We investigate the sorption behavior of Pu(IV) to FEBEX bentonite across a wide range of initial

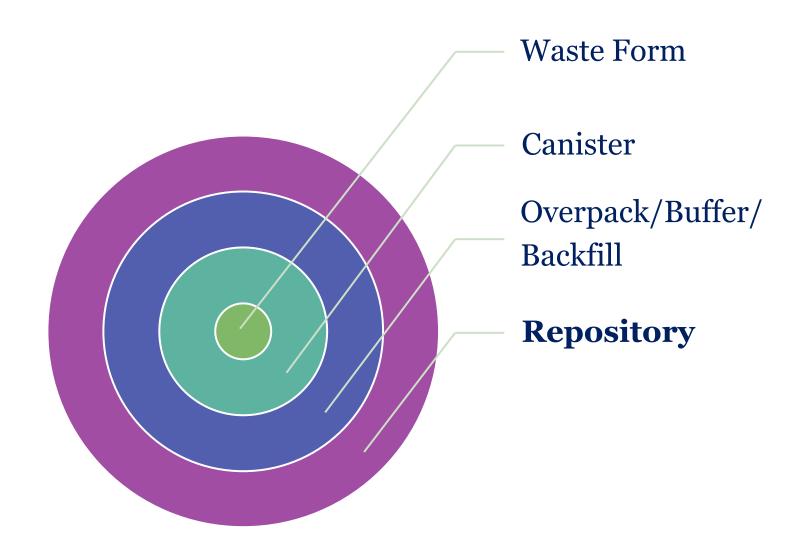
Clay backfill

"After vault backfilling, retrieval of the waste packages would still be possible [16], but additional equipment would be required and the retrieval would be more costly. Even after the repository is closed and sealed, the waste could be retrieved by conventional mining techniques."

Retrieval process for UK's NIREX concept

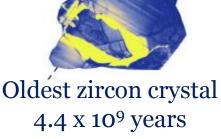


Multi-Barrier Approach

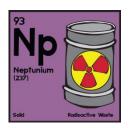


Why a geologic repository?





Np-237 half life 2.1 x 10⁶ years



4.4 x 10⁹ years Volcanic Tuff (USA) – last significant eruption **12** x **10**⁶ years

'S

Origin of Earth 4.6 x 10⁹ years

Clay rock (France, Switzerland)

Rock salt dome (Germany)

Granite Scandinavian shield (Finland/Sweden) - 1.2-3 x 109 years

3+100



Australopithecus $5 \times 10^6 \text{ years}$

Mol (clay)

None

Pinawa (granite)*

Bure (argillite)

Gorleben (salt)

(shallow)

Aspo (granite)

None

None

Exploratory studies facility at YM served

function of underground laboratory (tuff)

Construction of ONKALO underground

Tona (granite), Mizunami (granite),

Korea Underground Research Tunnel

More Terri (clay), Grirael (granite)

Horonaobe (sedimentary rock)

facility in Eurajoki began in 2004 (granite)

Salt, basalt, granite, tuff,

Granite, sedimentary

grandiorite, migmatite

Granite, sedimentary

Granite, clay, salt

No decision made.

Granite, gneiss,

Argillite, granite

clay, shale

Clay, shale

Granite

Salt

Granite

Granite

Clay, granite

US

Belgium

Canada

China

Finland

France

Japan

Korea

Spain

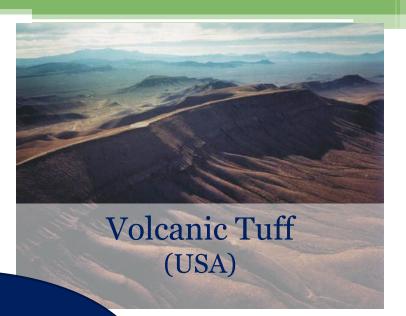
UK

Sweden

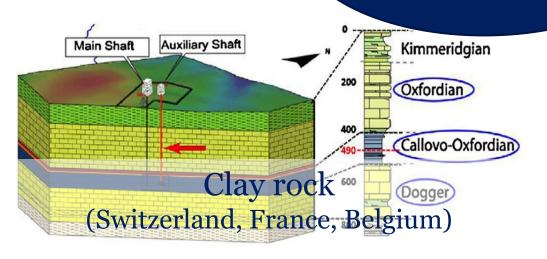
Switzerland

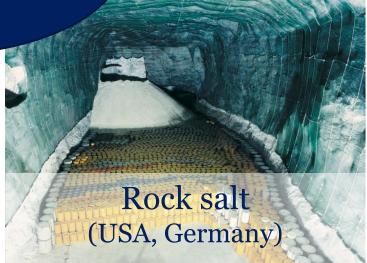
Germany





GEOLOGIC FORMATIONS





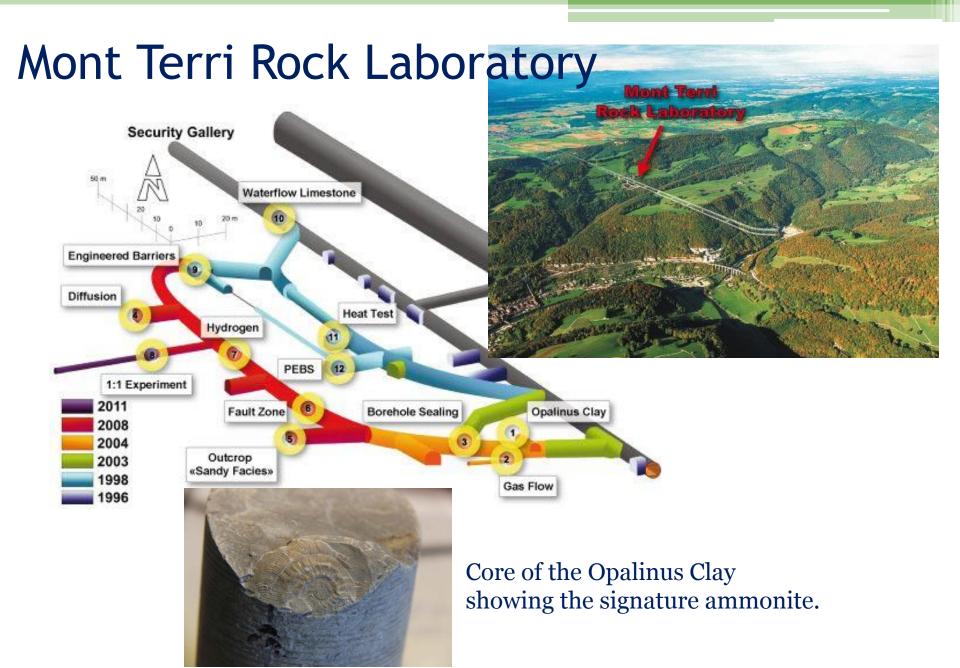
Clay Rock

Advantages

- Tightness
- Plasticity (swelling capacity)
- Low solubility
- High sorption capacity

Disadvantages

- Low heat conductivity
- Low temp. resistance
- Difficult mine construction
 - Damage zone around excavation
 - Oil drill holes common



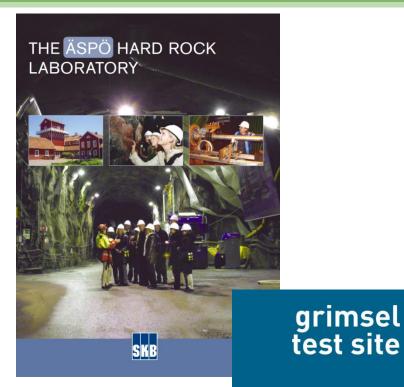
Granite

Advantages

- Mechanically stable
- Age of rock formation
- Moderate heat conductivity
- Good state of knowledge

Disadvantages

- Water bearing fractures
- Moderate retention capacity
- Technical barriers imperative (bentonite, copper canister)
 - Low temperature resistance



August 2004



ONKALO underground rock characterization facility (Finland)



Summer 2014

ANTO C

research on safe geological disposal of radioactive waste

nagra



Pinawa underground facility (Canada)

1983 - Construction begins

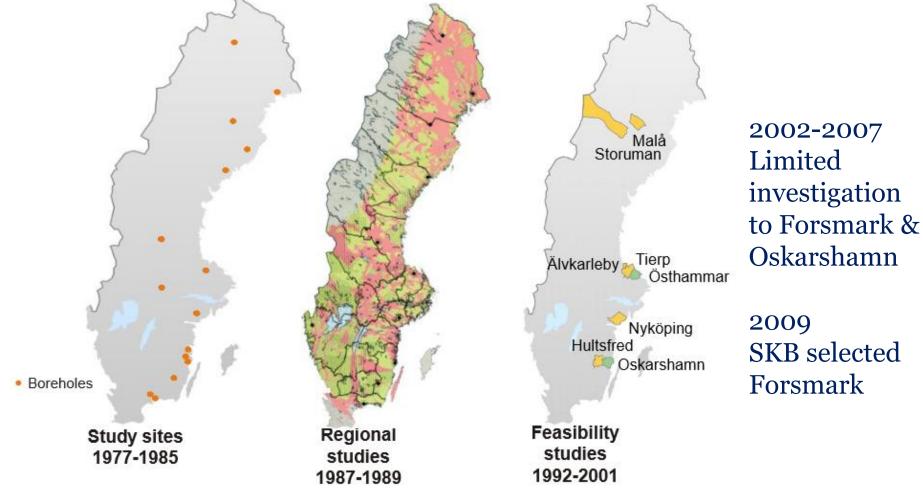
1985 – Research begins

1998 – Decommissioning begins

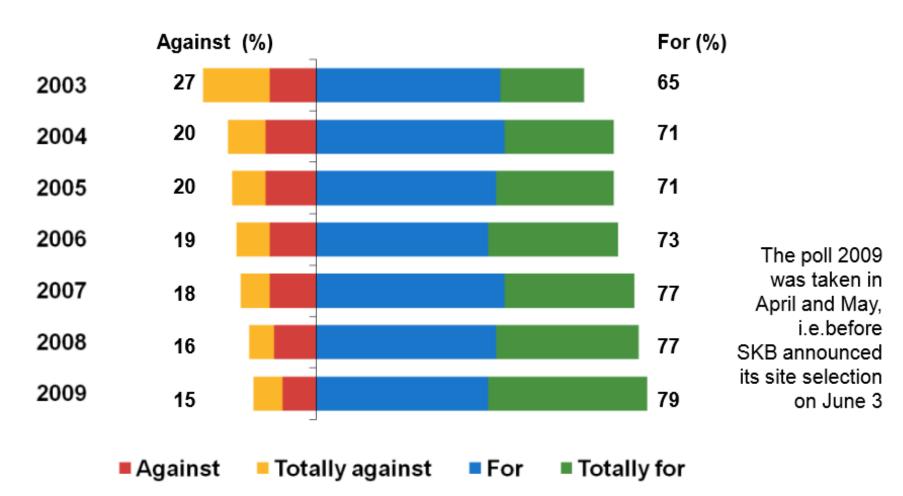
2010 – Final closure



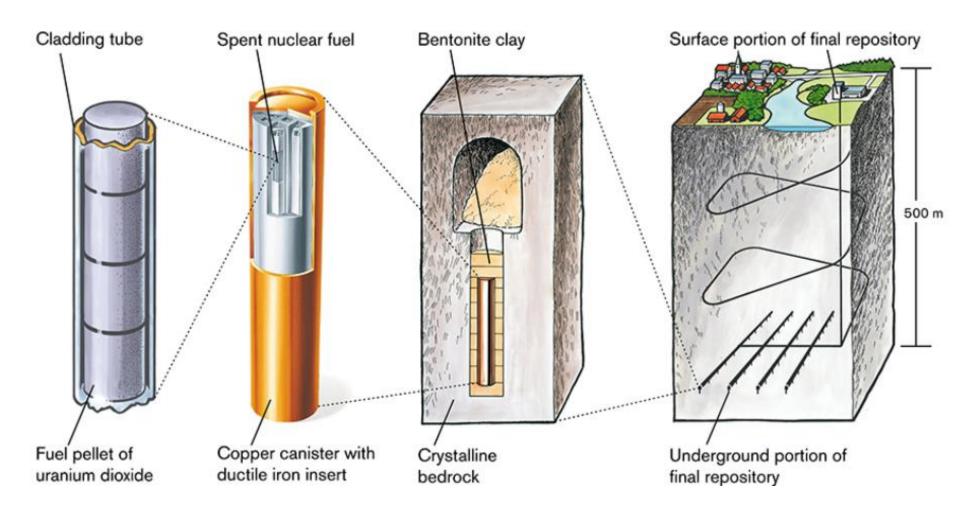
Sweden's Hard Rock Adventure: Siting studies (1977-2001)



Sweden's Hard Rock Adventure: Strong local support for the Forsmark



Sweden's Hard Rock Adventure: The KBS-3 design



Rock salt

Advantages

- Tightness
- Plasticity (convergence)
- Heat conductive
- High temp. resistance
- Age of existing diapirs
- Good state of knowledge

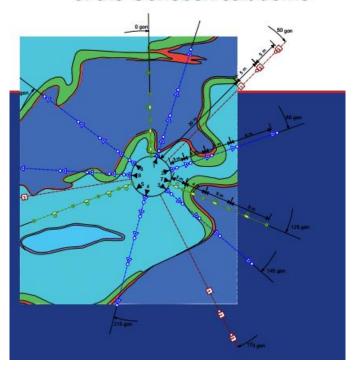
Disadvantages

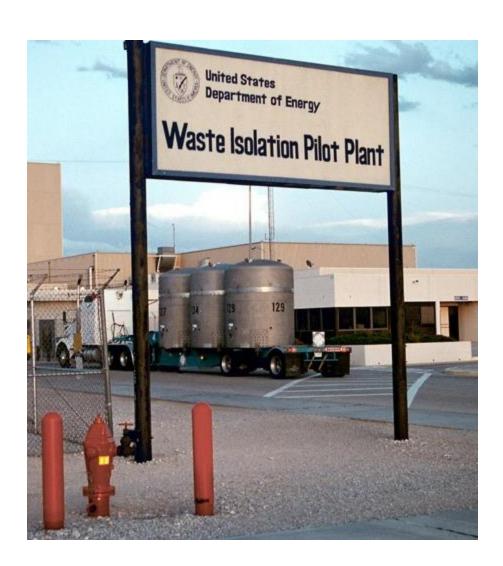
- Water soluble
- Low retention capacity
- Dissolution
- Uplift (~ 0.02 mm/yr)



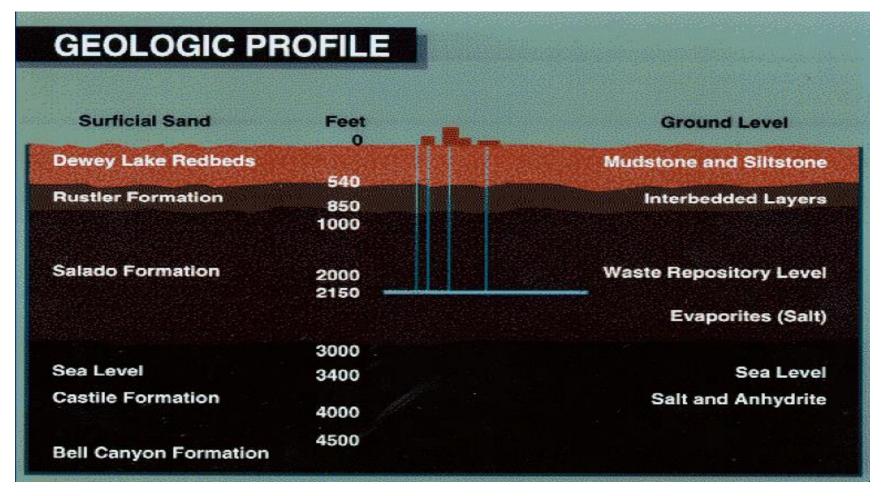
Description of the Gorleben site Part 4:

Geotechnical exploration of the Gorleben salt dome





Salt formations at WIPP were deposited in thick beds during the evaporation of the Permian Sea.



Hydrocarbons, potash, and possibly natural gas exist under WIPP or in the area.

March 26, 1999 - waste disposal operations begin



http://www.wipp.energy.gov/Video/w1.mpg

Shipments Received (as of February 11, 2014)

Site	Shipments	Loaded Miles
ANL	193	331,333
Bettis Atomic Power Lab	5	10,955
GE Vallecitos Nuclear Center	32	44,800
INL	5,844	8,132,064
LANL	1,344	459,648
LLNL	18	24,804
Nevada Test Site	48	57,312
ORNL	131	175,933
Rocky Flats	2,045	1,446,444
Hanford Site	572	1,034,176
SNL	8	2,200
Savannah River Site	1,654	2,483,360
Total to WIPP	11,894	14,203,029



Waste Isolation Pilot Plant WIPP Status Report As of 03/09/13

REPORT, WEEKLY March 13, 2013 7:32 AM Page 2 of 3

SHIPMENTS and VOLUME RECEIVED AT WIPP

Site	Last Week (02/24/13- 03/02/13)	Current Week (03/03/13- 03/09/13)	Total Shipments Rec'd to date: 03/09/13	Total Volume (m^3) Emplaced to date: 03/09/13	FY 2013 Vol (m^3) Emplaced to date: 03/09/13
ARGONNE NATIONAL LABORATORY - EAST - CH	0	0	14	120.78	0.00
ARGONNE NATIONAL LABORATORY - EAST - RH	0	0	142	53.94	3.97
BETTIS ATOMIC POWER LABORATORY - RH	0	0	5	3.15	0.00
GE VALLECITOS NUCLEAR CENTER - RH	0	0	32	19.74	0.00
HANFORD SITE - CH	0	0	572	5,060.79	0.00
IDAHO NATIONAL LABORATORY - CH	0	0	5,157	40,151.67	646.64
IDAHO NATIONAL LABORATORY - RH	1	2	313	146.46	9.66
LAWRENCE LIVERMORE NATIONAL LABORATORY - CH	0	0	18	146.14	0.00
LOS ALAMOS NATIONAL LABORATORY - CH	0	5	1,127	7,550.40	407.23
LOS ALAMOS NATIONAL LABORATORY - RH	0	0	16	14.24	0.00
NEVADA TEST SITE - CH	0	0	48	405.37	0.00
OAK RIDGE NATIONAL LABORATORY - CH	0	0	58	414.52	0.00
OAK RIDGE NATIONAL LABORATORY - RH	0	0	73	45.99	0.00
ROCKY FLATS ENVIRONMENTAL TECHNOLOGY SITE - CH	0	0	2,045	15,061.94	0.00
SANDIA NATIONAL LABORATORIES/NM - RH	0	0	8	4.62	0.00
SAVANNAH RIVER SITE - CH	0	1	1,450	16,299.89	673.86
SAVANNAH RIVER SITE - RH	0	0	43	25.35	0.00
WASTE ISOLATION PILOT PLANT - CH	0	0	0	3.90	0.00
Totals:	1	8	11,121	85,528.89	1,741.36



Waste Isolation Pilot Plant WIPP Status Report As of 03/09/13

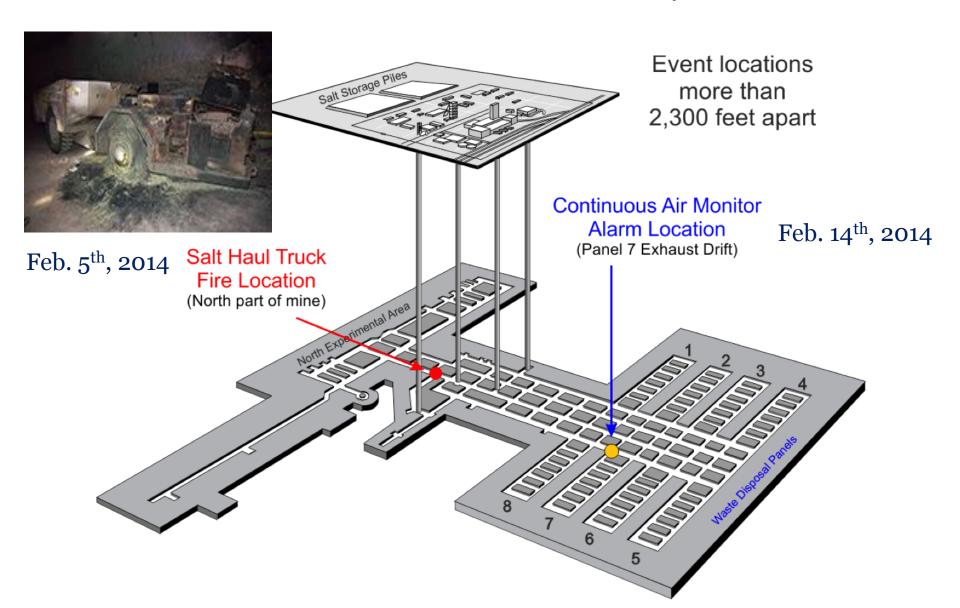
REPORT, WEEKLY March 13, 2013 7:32 AM Page 3 of 3

REPOSITORY

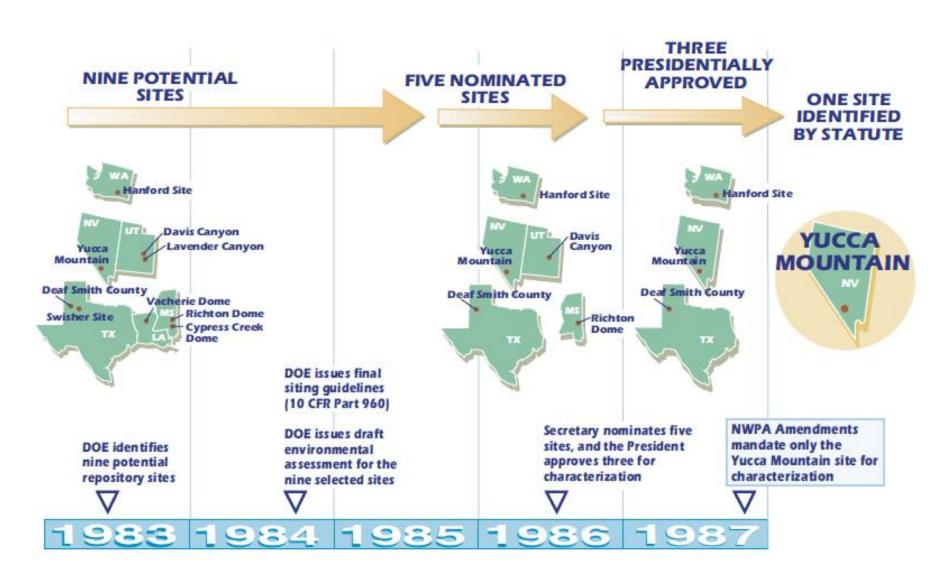
	Panel 1	Panel 2	Panel 3	Panel 4	Panel 5	Panel 6	
Emplaced Waste	CLOSED	CLOSED	CLOSED	CLOSED	CLOSED	ACTIVE	Total
# of 55-GALLON DRUMS	38,139	23,865	8,394	12,858	21,255	10,071	114,582
# of STANDARD WASTE BOXES	1,239	3,176	1,730	1,405	2,200	2,097	11,847
# of TEN DRUM OVERPACKS	35	1,451	2,227	1,048	788	206	5,755
# of 85-GALLON DRUM - TALLS	2	0	0	3	0	0	5
# of 100-GALLON DRUMS	0	1,278	5,409	11,050	9,951	4,587	32,275
# of STANDARD LARGE BOX 2S	0	0	0	0	0	97	97
# of REMOVABLE-LID 72-B CANISTERS	0	0	0	198	246	165	609
# of FIXED-LID 72-B CANISTERS	0	0	0	0	18	0	18
CH container volume (m^3)	10,496.65	17,997.67	17,092.06	14,257.54	15,926.93	9,444.55	85,215.40
RH container volume (m^3)	0.00	0.00	0.00	84.24	153.37	75.88	313.49
Total Volume (m^3)	10,496.65	17,997.67	17,092.06	14,341.78	16,080.30	9,520.43	85,528.89

http://www.wipp.energy.gov/general/GenerateWippStatusReport.pdf

Waste Isolation Pilot Plant February 2014



What about civilian waste in the US?



What sets YM apart from all other repositories?

- Above water table → oxidizing
 - take advantage of heat generating waste to drive off water
 - capacity limited based on heat distribution
 - additional engineered barrier (Ti drip shield)

reducing

coffinite USiO₄

uraninite (UO_{2+x})



oxidizing

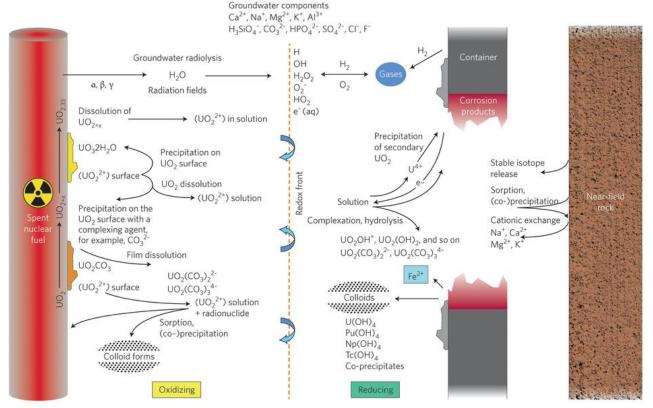
rutherfordine
metaschoepite becquerelite
autunite
sklodowskite studtite
uranophane weeksite
zippeite soddyite
torbernite clarkeite
schoepite
boltwoodite

The intersect of science and policy \rightarrow makes nuclear waste disposal even more challenging

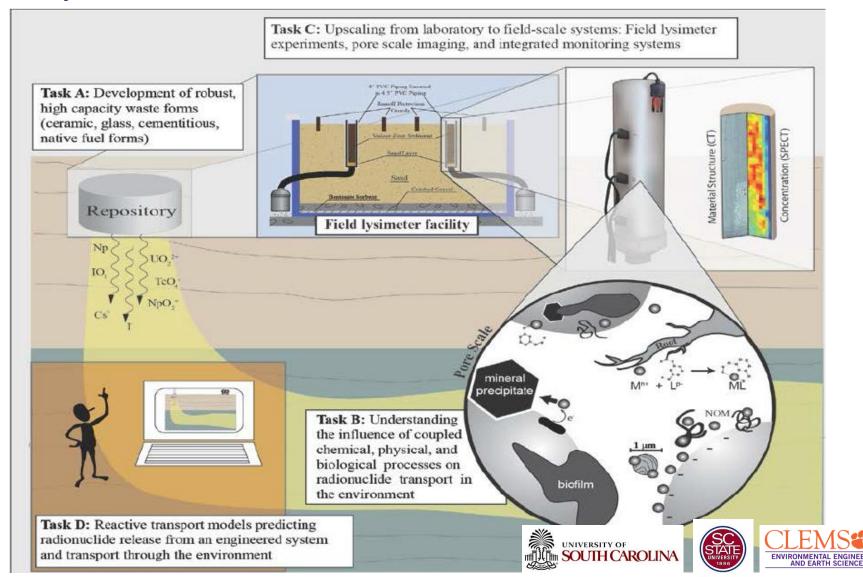
- Some political/societal challenges that impact repository design:
 - Location
 - Retreivability
 - Nonproliferation/Safeguards
 - Safety (in terms of accidental or eventual release)
- Educated decisions require sound scientific basis

We still need to uncover many scientific uncertainties...

Knowledge gaps remain that make predicting the fate and transport of RNs in the near field and far field of a repository challenging



Integrated science/engineering approach to fate and transport of RN in the environment



References

- Linking Legacies (1997) DOE-EM
- Ewing, R.C. (2015) "Long-term storage of spent nuclear fuel" *Nature Materials* 14: 252-257.
- Government Accountability Office (2014) Report GAO-15-141
- National Academy of Sciences (1957) "The Disposal of Radioactive Waste on Land" Report of the Committee on Waste Disposal of the Division of Earth Sciences
- Blue Ribbon Commission on America's Nuclear Future (2012) "Report to the Secretary of Energy"
- Hedin, A. (1997) "Spent Nuclear Fuel How Dangerous Is It?" SKB Technical Report 97-13, Swedish Nuclear Fuel and Waste Management Co., 60 pp.
- Bruno, J. and Ewing, R.C. (2006) "Spent Nuclear Fuel" Elements, 2: 343-349.
- www.world-nuclear.org
- Schneider, M. and Marignac, Y. (2008) "Spent Nuclear Fuel Reprocessing in France" International Panel on Fissile Materials Research Report No. 4.
- U.S. Nuclear Waste Technical Review Board (2009) "Survey of National Programs for Managing High-Level Radioactive Waste and Spent Nuclear Fuel" A Report to Congress and the Secretary of Energy.
- Civilian Radioactive Waste Management System Management & Operating Contractor (2000) "Total system Performance Assessment for the Site Recommendation" TDR-WIS-PA-000001 REV 00 ICN 01.
- Grambow, B. (2006) "Nuclear Waste Glasses How Durable?" Elements 2: 357-364.
- Lumpkin, G.R. (2006) "Ceramic Waste Forms for Actinides" Elements 2: 365-372.
- Wicks, G.G., Lodding, A.R., and Molecke, M.A. (1993) "Aqueous Alteration of Nuclear Waste Glasses and Metal Package Components" Materials Research Society Bulletin 18(9): 32-39.
- Grambow, B. (1994) "Borosilicate Glass: Future Research Requirements or 'What We Don't Know'" Materials Research Society Bulletin 19(12): 20-23.
- Palenik, C.S., Nasdala, L., and Ewing, R.C. (2003) "Radiation damage in zircon" American Mineralogist 88: 770-781.
- Weber, W.J. (2012) "Fundamental Aspects of Radiation Effects in Ceramics" Presentation for the EFRC Summer School 2012: Defects, Deformation and Damage in Structural Materials.
- Weber, W.J., Ewing, R.C., and Wang, L. (1994) "The radiation-induced crystalline-to-amorphous transition in zircon" Journal of Materials Research 9: 688-698.

References

- Kostya Trachenko, Queen Mary University, http://homes.esc.cam.ac.uk/kot/radiation-damage.
- O'Loughlin, E.J., Kelly, S.D., Cook, R.E., Csencsits, R., and Kemner, K.M. (2003) "Reduction of Uranium(VI) by Mixed Iron(II)/Iron(III) Hydroxide (Green Rust): Formation of UO₂ Nanoparticles" Environmental Science and Technology 37: 721-727.
- Christiansen, B.C., Geckeis, H., Marquardt, C.M., Bauer, A., Romer, J., Wiss, T., Schild, D., Stipp, S.L.S. (2011) "Neptunyl (NpO₂⁺) interaction with green rust, GRNa,SO₄" Geochimica et Cosmochimica Acta 75: 1216-1226.
- Ferriss, E.D.A., Helean, K.B., Bryan, C.R., Brady, P.V., and Ewing, R.C. (2009) "UO₂ corrosion in an iron waste package" Journal of Nuclear Materials 384: 130-139.
- Puigdomenech et al. "Thermodynamic data for copper: Implications for the corrosion of copper under repository conditions" SKB 2000.
- Renock, D. and Shuller-Nickles, L. (2015) "Predicting Geologic Corrosion with Electrodes" *Elements* accepted.
- Couture, R. (1984) "Some issues on the use of backfill materials in high-level nuclear waste repositories" ANL draft report http://pbadupws.nrc.gov/docs/ML0320/ML032060230.pdf.
- Begg J.D. et al. *J. of Env. Radioact. 141 (2015) 106.*
- IAEA (20) "Geological Disposal of Radioactive Waste: Technological Implications for Retrievability" IAEA Nuclear Energy Series No. NW-T-1.19
- http://www.mont-terri.ch/internet/mont-terri/en/home/rock_lab/view_from_outside.html
- http://www.winnipegfreepress.com/breakingnews/whiteshell-labs-closes-underground-facility-forever-111511344.html
- http://www.grimsel.com/images/stories/pdfs/e_flg10.pdf
- http://www.posiva.fi/en/final_disposal/onkalo/the_construction_of_onkalo#.VXdIMc9VhBc
- http://www.skb.se/upload/publications/pdf/Aspo_Laboratory.pdf
- https://ec.europa.eu/jrc/sites/default/files/jrc aaas2010 waste thegerstrom.pdf
- www.wipp.energy.gov
- http://www.wipp.energy.gov/general/GenerateWippStatusReport.pdf

References for Figures/Videos

- Slide 2 http://iss.unige.ch/content/phd-thesis-defense-nizar-ghoula;;
 http://co2.egi.utah.edu/
- Slide 5 www.nei.org
- Slide 7 https://commons.wikimedia.org/wiki/File:Saqqara pyramid

 ver 2.jpg
- Slide 34 https://www.youtube.com/watch?v=1eJMY9MT4a8&feature=youtu.be&t=11
- Slide 44 modified from H. Gekeis
- Slides 51-52 modified from C. Thegerström
- Slide 53 http://www.skb.com/future-projects/the-spent-fuel-repository/our-methodology/
- Slide 57 http://www.wipp.energy.gov/Video/w1.mpg
- Slide 66 Clemson University EPSCoR Implementation grant

Upcoming Webinars

- High Level Waste
- •High Resolution Gamma-Ray Spectrometry Analyses for Normal Operation and Radiological Incident Response
- Nuclear Radiation Safety

NAMP website: www.wipp.energy.gov/namp